KABARAK



UNIVERSITY

EXAMINATIONS

2008/2009 ACADEMIC YEAR

FOR THE DEGREE OF BACHELOR OF EDUCATION SCIENCE

- COURSE CODE: CHEM 312
- COURSE TITLE: ORGANIC CHEMISTRY III
- STREAM: SESSION V & VII
- DAY: WEDNESDAY
- TIME: 9.00 11.00 A.M.
- DATE: 08/04/2009

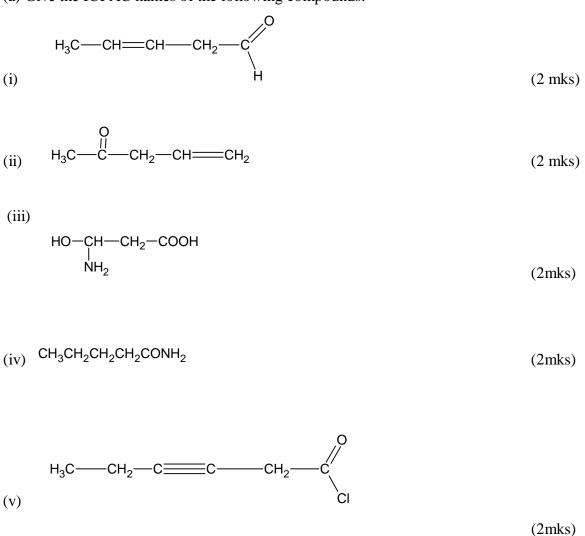
INSTRUCTIONS:

Attempt all questions

PLEASE TURN OVER

QUESTION ONE 17 1/2 MKS

(a) Give the IUPAC names of the following compounds:

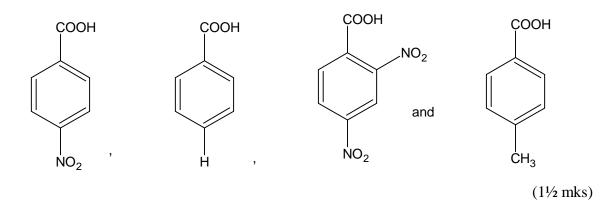


- b) Describe a simple chemical test that would distinguish between each of the following pairs of compounds.
 - i) Pentane and 3- pentanone
 - ii) Benzoic acid and Benzyl alcohol

{4mks}

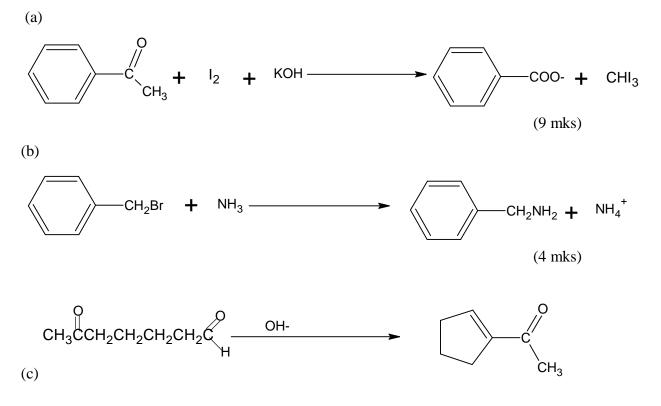
c) Briefly explain why aldehydes and ketones generally have lower boiling points than alcohols of comparable molecular size.
{2mks}

d) Arrange the following compounds according to their increasing acidity.



QUESTION TWO - 17 ¹/₂ mks

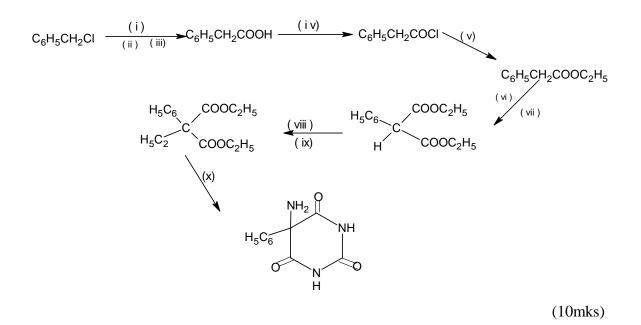
Write a possible mechanism for each of the following reactions



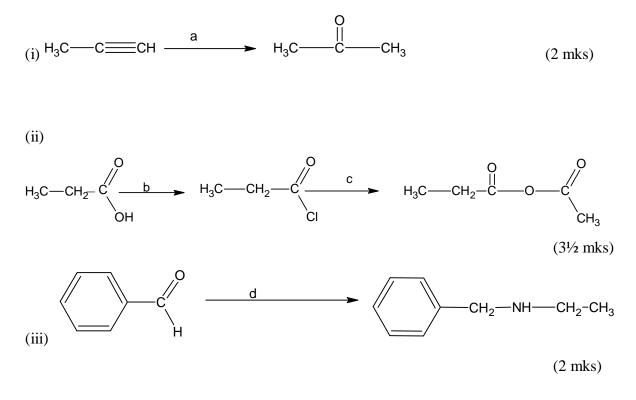
(4 ½ mks)

QUESTION THREE 17 1/2 MKS

a) Outlined below is a synthesis of phenolbarbital identify the reagents required for each transformation;



b) Give the reagents required to carry out each of the following reactions.

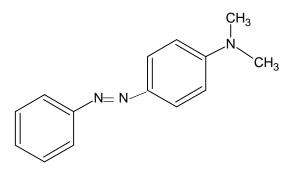


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QUESTION FOUR 17 ¹/₂ MKS.

a) Butter yellow is a dye which was once used to colour Margarine. It has since been shown to be carcinogenic and its use in food is no longer permitted. Outline a synthesis of butter yellow from Benzene and N, N – Dimethylaniline.

(8 ½ mks)



(b) Phenacetin is an analgesic and antipyretic compound, and was the P of APC tablets (Asprin – Phenacetin – Caffeine) Phenacetin is no longer used medically due to its toxicity. The following is the synthesis of phenacetin, propose structures for the intermediates A – C and phenacetin.

4- Ethoxyaniline
$$\begin{array}{c} 1. \text{HONO, H}_{3}\text{O}^{+} \\ \hline 2. \text{Phenol, OH-} \end{array} A.(C_{14}\text{H}_{14}\text{N}_{2}\text{O}_{2}) \\ \hline CH_{3}\text{CH}_{2}\text{Br} \end{array} \xrightarrow{B.(C_{16}\text{H}_{18}\text{N}_{2}\text{O}_{2})} \\ \hline SnCl_{2} \\ \hline \\ (C_{10}\text{H}_{13}\text{NO}_{2}) \text{Phenacetin} \end{array} \xrightarrow{Acetic anhydride} 2 \text{ Moles C. } (C_{8}\text{H}_{11}\text{NO}) \end{array}$$

(9 mks)