



UNIVERSITY

UNIVERSITY EXAMINATIONS

2008/2009 ACADEMIC YEAR

FOR THE CERTIFICATE OF PRE- UNIVERSITY CHEMISTRY

COURSE CODE: PCHEM 021

COURSE TITLE: INTRODUCTION TO ORGANIC

CHEMISTRY

STREAM: SEMESTER TWO

DAY: TUESDAY

TIME: 9.00 - 11.00 A.M.

DATE: 11/08/2009

INSTRUCTIONS:

Attempt all questions.

PLEASE TURN OVER

QUESTION ONE (20 MARKS)

- a) Define the following terms.
 - i. Functional group
 - ii. Hydrocarbons
 - iii. Isomers (6marks)
- b) Differentiate between the following;
 - i. Saturated hydrocarbons and unsaturated hydrocarbons
 - ii. Cyclic and acyclic hydrocarbons

(4marks)

- c) Enumerate three unique properties of carbon that enables it form a wide variety of compounds unlike other atoms. (2marks)
- d) Draw the possible isomers of C_6H_{14} .

(8marks)

QUESTION TWO (20 MARKS)

a) Assign the IUPAC names to the following structures.

iii.

(10marks)

- b) Give structures to the following IUPAC names of compounds.
 - i. Propyne

iv. 2-methyl-2-butene

ii. 2-methylpentane

v. 1,3,5-hexatriene

iii. Trichloroethene

(10marks)

QUESTION THREE (15 MARKS)

- a) Explain the following observations
 - i. Alcohols are more soluble in water than corresponding alkanes. (3marks)
 - ii. The boiling points of branched alkanes are lower than those of corresponding straight chain alkanes. (2marks)
 - iii. Alkenes are more reactive than alkanes. (2marks)
- b) Explain the nature of the C=C double bond. (4marks)
- c) Give the functional groups corresponding to the following groups of organic compounds.
 - i. Ketones
 - ii. Esters
 - iii. Carboxylic acids
 - iv. Ethers (4marks)

QUESTION FOUR (15MARKS)

iii.

a) Complete the following reactions.

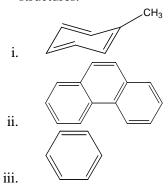
$$\begin{array}{ccc} \text{CH}_3\text{CH}_2\text{CH}_3 & \xrightarrow{\hspace{1cm} \text{O}_2 \\ \end{array}} \\ \text{i.} & \text{Heat} \end{array}$$

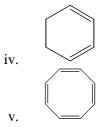
ii. H_2C — CHC H_2CH_3 + HCl —

+ Cl₂ FeCl₃

iv. FeCl₃

b) Identify which of the following compounds are aromatic and non-aromatic out of the following structures.





(5marks)