Date: DECEMBER 2015

## THE CATHOLIC UNIVERSITY OF EASTERN AFRICA

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**Duration: 2 Hours** 

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### MAIN EXAMINATION

# AUGUST - DECEMBER 2015 TRIMESTER

### **FACULTY OF SCIENCE**

### DEPARTMENT OF CHEMISTRY

### SCHOOL FOCUSED PROGRAMME

CHEM 310: ENVIRONMENTAL CHEMISTRY

INSTRUCTIONS: Answer Question ONE and any other TWO Questions Q1. a) Explain green house effect causes global environmental impact and how it can be minimized. (8 marks) b) Explain why supersonic jets are a threat to the environment. (3 marks) c) Define the following and give examples where applicable. Dissolved oxygen ii **Environmental pollution** iii Hydrosphere İ۷ Waste management Chemical Oxygen Demand (5 marks) d) Write short notes on the following i Global warming (4 marks) ii Environmental impacts of acid rain. (5 marks) Emmission from automobiles and their effects. iii (5 marks) Q2. a) Sulphur dioxide (SO<sub>x</sub>) are serious pollutants to the environment. Discuss. (15 marks) b) Define entrophication and show how it arises. (5 marks)

- Q3. a) Explain the environmental impact of solid wastes and ways of removing solid waste from the environment. (3 marks)
  - b) Why is excessive use of fertilizers an environmental hazard. (3 marks)
  - c) State THREE classes of pesticides. (3 marks)
  - d) Give an example of polyaromatic hydrocarbons PAHs and explain how environmental pollution from these compounds can be minimized.
    (4 marks)
- Q4. a) Lead and Cadmium are toxic heavy metals. Explain their sources in the environment, effect on human health and ways of reducing their environmental impact. (10 marks)
  - b) Ozone is an important component of the stratosphere whose levels keep fluctuating due to anthropogenic activities. Discuss ozone in terms of
    - i Formation mechanism
    - ii Depletion mechanism
    - iii Effects depletion

(10 marks)

- Q5. a) Acid rain is mainly formed from NO<sub>x</sub> and SO<sub>x</sub> describe how levels of NO<sub>x</sub> and SO<sub>x</sub> can be determined in the environment. (10 marks)
  - b) Whereas N<sub>2</sub>O is essentially inert in the troposphere it is a major pollutant in the stratosphere. Discuss this observation. (3 marks)
  - c) Discuss various control measures of SO<sub>x</sub> in the atmosphere. (4 marks)
  - d) Briefly explain why the stratosphere is more susceptible chemical pollution than the troposphere. (3 marks)

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