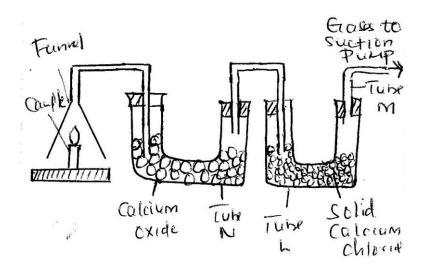
## KIPSUTER BOYS SECONDARY SCHOOL

## **CAT 1 TERM 2, 2017**

FORM '	TWO	HEMISTRY			
Name: _		Adm.	. No	Class:	Date:
Instruct	ions:				
i	i). Write your name, admission n	umber, class & date	in the spaces pro	ovided above.	
i	ii). Check the question paper to a	scertain that all page	es are printed as	indicated and	that no question
	is missing.				
i	iii). Answer <b>ALL</b> questions in the s	spaces provided after	r each question.		
1. 8	a) The grid given below represe	ents part of the perio	odic table. Stud	ly it and answ	ver the
(	questions that follow. (The lette	ers do not represent	the actual sym	bols of the el	ements).
	i). What name is given to to	he group of elemen			mk)
	iii). What is the formula of t	he compound form	ed when eleme	nt D and oxy	gen gas react.
	iv). What name is given to t	ne group of elemen	ts to which M a	and G belong	s?(1mk)
	v). To what period does D l	pelongs?(1mk)			

- vi). On the grid, indicate with a tick the position of element x which is in the third period of the periodic table and forms  $x^{3-}$  ions. (1mk)
- b) State three uses of the elements in group O (Noble gases). (3mks)

- 2. a) Candle wax is mainly a compound consisting of two elements. Name the two elements.(2mks)
  - b) The set-up below was used to investigate the burning of a candle. Study it and answer the questions that follow.



i). What would happen to the burning candle if the pump was turned off? (3mks)

ii). State and explain the changes in the mass that are likely to occur in tube N by the end of the experiment. (3mks)

	iii). Name two gases that come out through tube M. (2mks)
	iv). What is the purpose of calcium chloride in tube L?(1mk)
	v). Name another substance that could be used in place of calcium oxide in tube N. (1mk)
3.	a)What is meant by the terms: (2mks)  i). Element
	ii). Atomic number
	b) If the formula for a chloride of titanium is $TiCl_3$ . What is the formula of its sulphate? (1mk)
	c) What are isotopes? (1mk)
	d) Determine the number of neutrons in ${}^{18}_{8}O$ (1mk)
4.	Distinguish between ionization energy and electron affinity. (2mks)
5.	Giving one example, define the term radical. (2mks)