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UNIVERSITY OF EMBU

2017/2018 ACADEMIC YEAR

TRIMESTER EXAMINATION

SECOND YEAR EXAMINATION FOR THE DEGREE OF BACHELOR OF SCIENCE (NURSING)

HNS 133: MEDICAL BIOCHEMISTRY II

DATE: JULY 31, 2018		TIME: 11:00-2:00PM	
Ar Al Al Ca	NSTRUCTIONS: nswer: II MCQs in Section A; II Short-answer Questions in Section B II Long-answer Questions in Section C ancelled work should be done neatly by crossing the MCQs	ng with a single line in the essay and by use	
SE	ECTION A: MULTIPLE CHOICE QUESTIO	NS (TOTAL: 20 MARKS)	
Cł	hoose (CIRCLE/TICK) only one correct answ	er from the following questions	
1.	Vitamin D is	ē	
	a) Not a hormone, it is an element consumed i	n the diet.	
	b) A peptide hormone that must be processed	in the body into a biologically active form.	
	c) A steroid hormone which must be processed	d in the body into a biologically active form.	
	d) A steroid hormone which does not need to	o be processed by the body into a biologically	
	active form.		
2.	Endocrine hormones		
	a) Are released by ductless glands and act on	nearby cells via diffusion.	
	b) Are released into the bloodstream for action	n on receptor cells distant from the source.	
	c) Are released only by the adrenal glands.		

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Knowledge Transforms

	d)	bind to special proteins which enable them to act on the DNA.
3.	Hi	gh levels of dopamine are implicated in
	a)	Parkinson disease
	b)	Attention deficit disorder.
	c)	Schizophrenia.
	d)	Alzheimer disease.
4.	Co	ortisol can be characterized as a hormone which
	a)	Increases blood pressure.
	b)	Increases vasoconstriction, but has minimal effect on blood pressure.
	c)	Reduces blood pressure.
	d)	Is only involved in stressful effects on the immune system
5.	Th	e final intermediate product in the citric acid cycle is
	a)	L-malate.
	b)	Acetyl-CoA.
	c)	Oxaloacetate.
	d)	Fumarate.
6.	Isc	ocitrate dehydrogenase
	a)	Is activated by high concentrations of ATP and NADH.
	b)	Is activated by high concentrations of ATP and NADPH.
	c)	Is unaffected by high concentrations of NADPH.
	d)	Is inhibited by high concentrations of high-energy compounds.
7.	In	the electron transport chain, complex II
	a)	Does not contribute to the proton gradient, but mediates the transfer of electrons from
		succinate to cytochrome a.
	b)	Does not contribute to the proton gradient, but mediates the transfer of electrons from
		succinate to cytochrome c.
	c)	Does not contribute to the proton gradient, but mediates the transfer of electrons from
		succinate to coenzyme Q.
	d)	Contributes to the proton gradient, and mediates the transfer of electrons from succinate to
		coenzyme

8. Gluconeogenesis o		uconeogenesis occurs in organelles called	
		a) Spherosomes	
		b) Lysosomes	
		c) Mitochondria	
		d) Glyoxysomes	
9.)is the terminal electron acceptor during oxidative phosphorylation.		
		a) FADH	
		b) NAD	
		c) Oxygen	
		d) ATP	
10.	Th	e formation of a nucleic acid involves	
	a)	Adehydration process involving a base, a phosphate, and a pentose sugar.	
	b)	A hydrolysis process involving a base, a phosphate, and a pentose sugar.	
	c)	A dehydration process involving a base, a phosphate, and a hexose sugar.	
	d)	None of the above.	
11.	Pu	rine bases include:	
	a)	Adenine and cytosine	
	b)	Adenine and uracil	
c) Adenine and guanine		Adenine and guanine	
	d)	Adenine and thymine	
12.	DN	VA differs from RNA in some features such as:	
	a)	DNA residues are linked by 3'>5' phosphodiester bonds; RNA is 2'->5' linked.	
	b)	DNA has deoxyribose residues; RNA has ribose residues.	
	c)	DNA contains the A, C, G and T bases; RNA contains A, C, G, and U.	
	d)	All of the above	
13.	Gli	uconeogenesis	
	a)	Splits glucose to pyruvate which can be converted to lactate under anaerobic condition.	
	b)	Converts pyruvate to glucose.	
	c)	Synthesis of glycogen, carbohydrate fuel storage form.	
	d)	Produces NADPH for cell biosynthesis.	

14.	In	a eukaryotic cell, most of the enzymes of the citric acid cycle are located in
	the)
	a)	Mitochondrial matrix.
	b)	Inner mitochondrial membrane.
	c)	Intermembrane space of the mitochondria.
	d)	Outer mitochondrial membrane.
15.	Iso	ocitrate dehydrogenase
	a)	Is activated by high concentrations of ATP and NADH.
	b)	Is activated by high concentrations of ATP and NADPH.
	c)	Is unaffected by high concentrations of NADPH.
	d)	Is inhibited by high concentrations of high-energy compounds.
16.	Un	coupling agents of oxidative phosphorylation
	a)	Allow electron transport to continue but prevent the phosphorylation of ADP to ATP
	b)	Prevent electron transport from occurring but allow the phosphorylation of ADP to ATP
	c)	Block both electron transport and the phosphorylation of ADP to ATP
	d)	Are agents like rotenone and cyanide that block electron transport at specific carrier
17.	In	the pentose phosphate pathway, NADPH is produced at the step catalyzed by:
		a) Phosphopentose isomerase
		b) Gluconolactonase
		c) 6-phosphogluconate dehydrogenase
		d) Transketolase
18.	Th	e basic building blocks during biosynthesis of fatty acids are:
		a) Three-carbon units
		b) Two-carbon units
		c) Two-nitrogen units
		d) Glucose molecules
19.	Ch	olesterol
	a)	Is a precursor to most steroid hormones. The first step in the synthesis of a hormone
		involved the cleavage of a 6-carbon residue from cholesterol.
	b)	Is not involved in the synthesis of most steroid hormones. It is an undesirable component
		of the diet.

c) Is a precursor to most steroid hormones. The first step in the synthesis of a			sis of a normone		
		involved the addition of a 6-carbon residue from cholesterol.			
	d)	Is a precursor to all steroid hormones.			
20.	_	is the primary regulatory enzyme in fatty acid biosynthes	is.		
	a)	Acetyl-CoA carboxylase			
	b)	Pyruvate carboxylase			
	c)	Kinase			
	d)	Thiokinase			
SE	CT	ION B: Answer ALL the questions in this section (40 marks; Use th	e answer booklet		
pr	ovic	led)			
1.	Di	fferentiate transamination and reductive amination	(5 marks)		
2.	De	Describe the significance of the pentose phosphate pathway in the human body			
			(5 marks)		
3.	De	scribe the common symptoms of impaired production of gonadotropins	(5 marks)		
4.	Br	iefly describe five common urea cycle defects	(5 marks)		
5.	De	escribe the regulation of fatty acid oxidation	(5 marks)		
6.		scuss the general functions of hormones	(5 marks)		
7.	De	escribe the major causes of Ketosis	(5 marks)		
8.	Vi	tamins play a significant role in the TCA cycle. Explain	(5 marks)		
SE	CT	TION C: Answer ALL the questions in this section (40 marks; Use the	answer booklet		
pr	ovi	ded)			
1.	a) I	Describe the common areas of hormone production	(10 marks)		
b) Describe the different classes of hormones based on the distance they cover and chemical					
		composition.	(10 marks)		
2. a) Discuss the coupling concept and its significance in oxidative phosphorylation.					
	The state of the		(10 marks)		
	t	b) Discuss the uncoupling concept and its significance in oxidative phosph	orylation.		

(10 marks)

