



UNIVERSITY EXAMINATIONS 2018/2019 ACADEMIC YEAR

4th YEAR EXAMINATION FOR BACHELOR OF SCIENCE, BACHELOR OF SCIENCE (INDUSTRIAL CHEMISTRY) AND BACHELOR OF EDUCATION SCIENCE

(INSTITUTIONAL BASED PROGRAMME)

COURSE CODE/TITLE: SPH B203 MODERN PHYSICS

END OF SEMESTER II

DURATION: 2 HOURS

DAY/TIME: MONDAY: 9.00-11.00 AM DATE: 29/04/2019 (2ndF/EW)

INSTRUCTION: ANSWER ALL QUESTIONS.

QUESTION 1.

- (a) (i) Explain what is meant by the Uncertainty Principle. [3points]
- (ii) Consider a radio pulse lasting 0.0010 s. The position uncertainty of this pulse is $\Delta x = 3.0 \times 10^{-5} \text{ m}$. Determine the uncertainty in the momentum of the pulse and the uncertainty in the frequency of the pulse. [6points]
- (b) State two of Bohr's postulates. [3points]
- (c) From Bohr's theory of the atom, derive the expression for the Bohr radius of an atom. [6points]
- (d) Compute the value of the energy for an electron in the ground state

$$E = \frac{m_e e^4}{2(4\pi\epsilon_0)^2 \hbar^2 n^2}. \text{ Give your answer in electron volts. where } n = 1 \quad [7\text{points}]$$

QUESTION 2.

A particle of mass $m = 9 \times 10^{-31}$ kg is confined to move in a one dimensional region of length $0 < x < L$. Inside the region the potential is zero, and outside the potential is infinite.

(a) Write down the Hamiltonian of the particle if its charge is $e = 1.6 \times 10^{-19}$ C. [3points]

(b) Write down the Schrodinger equation for the particle. [4points]

(c) Given that the solution of the Schrodinger equation is,

$$\Psi(x) = N \sin(kx)$$

(i) Determine the normalization constant N . [6points]

(ii) Determine the value of k . [5points]

(iii) Compute the total energy of the particle in electron volts. [7points]

QUESTION 3.

(a) Explain the meaning of the following terms:

(i) Valence band and conduction band. [3points]

(ii) Rectifier diode. [3points]

(iii) Quantum well dot. [4points]

(b) (i) Explain what is meant by nuclear binding energy. [3points]

(ii) Compute the nuclear binding energy of the isotope ^{238}U . Express the energy in MeV. The mass of one atom of this isotope is 238.0508 u. [7points]

(Here $A = 238$ and $Z = 92$ $u = 1.6605 \times 10^{-27}$ kg and $c = 2.9979 \times 10^8 \frac{m}{s}$).